

2,3-Bis[(1-phenyl-1*H*-tetrazol-5-ylsulfanyl)-methyl]quinoxalineWei Wang,<sup>a\*</sup> Ning-Ning Pan<sup>a</sup>  
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## Key indicators

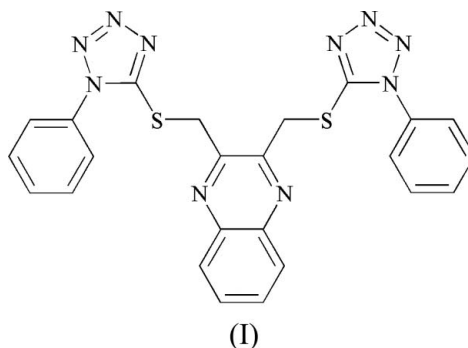
Single-crystal X-ray study  
 $T = 294$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.004$  Å  
 $R$  factor = 0.044  
 $wR$  factor = 0.133  
Data-to-parameter ratio = 14.9For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The title compound,  $\text{C}_{24}\text{H}_{18}\text{N}_{10}\text{S}_2$ , crystallizes with two independent molecules in the asymmetric unit. The two terminal (1-phenyltetrazol-5-yl)sulfanyl groups adopt a *trans* configuration with respect to the central quinoxaline ring system in both independent molecules.

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## Comment

To date, a large number of flexible or rigid chain-linked dithioether ligands containing *N*-heterocyclic groups have been synthesized and investigated, due to their diverse coordination capabilities and the important properties of their metal complexes (Zheng *et al.*, 2003; Bu *et al.*, 2002; Hong *et al.*, 2000). Earlier studies reported that several tetrazole derivatives possess varied pharmacological properties (Juby *et al.*, 1968, 1982). A new tetrazole derivative, the title compound, (I), has been synthesized and its crystal structure is presented here.



The asymmetric unit of compound (I) contains two independent molecules, *A* and *B*. The two terminal (1-phenyl-1,2,3,4-tetrazol-5-yl)sulfanyl groups adopt a *trans* configuration with respect to the central quinoxaline ring system in both molecules *A* and *B* (Fig. 1). In molecule *A*, the planar quinoxaline ring system makes dihedral angles of 11.5 (3) and 8.8 (3)° with the two tetrazolyl rings, indicating that the three groups are approximately coplanar. A similar result is also observed in molecule *B*. In molecule *A*, the two terminal phenyl rings make a dihedral angle of 39.5 (3)°, while in molecule, *B*, the dihedral angle is 19.2 (2)°.

Due to  $\pi$ - $\pi$  conjugation, the  $\text{Csp}^2$ -S bonds, *i.e.* S1-C1, S2-C18, S3-C25 and S4-C42, are significantly shorter than the  $\text{Csp}^3$ -S bonds, *i.e.* S1-C8, S2-C17, S3-C32 and S4-C41 (Table 1). The average lengths of the  $\text{Csp}^2$ -S and  $\text{Csp}^3$ -S bonds are 1.732 (3) and 1.807 (3) Å, respectively, which are comparable with those reported in the literature (Wang *et al.*, 2004, 2005).

## Experimental

A solution of 2,3-bis(bromomethyl)quinoxaline (1.58 g, 5 mmol) in ethanol (10 ml) was added dropwise to a mixture of 1-phenyl-5-thio-1,2,3,4-tetrazole (1.96 g, 11 mmol), KOH (0.62 g, 11 mmol) and ethanol (5 ml). The reaction mixture was then stirred for 24 h at room temperature. The precipitate which formed was filtered off, washed with water and recrystallized from ethanol (yield 70%, m.p. 446–447 K). Analysis, calculated for  $C_{24}H_{18}N_{10}S_2$ : C 56.45, H 3.55, N 27.43%; found: C 56.41, H 3.60, N 27.37%. Crystals of (I) suitable for single-crystal X-ray analysis were grown by slow evaporation of a solution in chloroform.

### Crystal data

$C_{24}H_{18}N_{10}S_2$	$V = 2345.8 (12) \text{ \AA}^3$
$M_r = 510.62$	$Z = 4$
Triclinic, $P\bar{1}$	$D_x = 1.446 \text{ Mg m}^{-3}$
$a = 6.3813 (18) \text{ \AA}$	Mo $K\alpha$ radiation
$b = 16.758 (5) \text{ \AA}$	$\mu = 0.26 \text{ mm}^{-1}$
$c = 22.214 (6) \text{ \AA}$	$T = 294 (2) \text{ K}$
$\alpha = 96.538 (5)^\circ$	Block, colourless
$\beta = 92.822 (4)^\circ$	$0.32 \times 0.20 \times 0.12 \text{ mm}$
$\gamma = 95.274 (5)^\circ$	

### Data collection

Bruker SMART CCD area-detector diffractometer	13253 measured reflections
$\varphi$ and $\omega$ scans	9645 independent reflections
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	4834 reflections with $I > 2\sigma(I)$
$T_{\min} = 0.910, T_{\max} = 0.969$	$R_{\text{int}} = 0.022$
	$\theta_{\max} = 26.5^\circ$

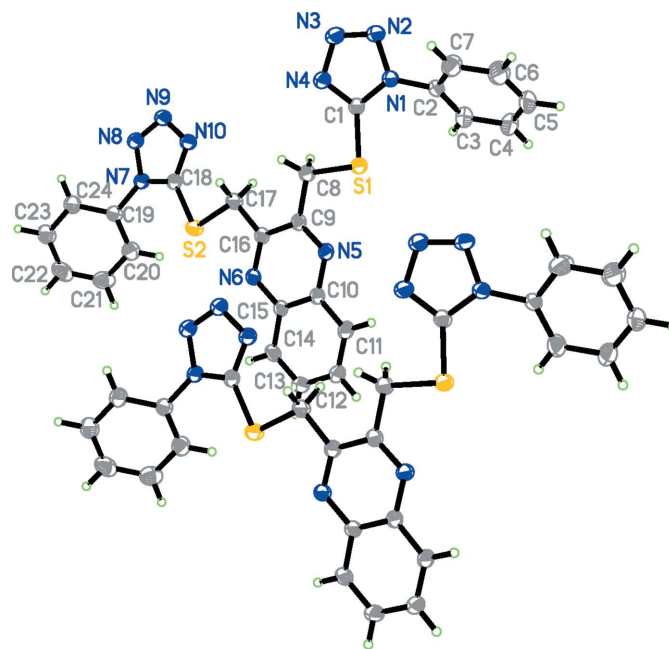
### Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.0533P)^2 + 0.326P]$
$R[F^2 > 2\sigma(F^2)] = 0.044$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.133$	$(\Delta/\sigma)_{\max} = 0.001$
$S = 0.99$	$\Delta\rho_{\max} = 0.27 \text{ e \AA}^{-3}$
9645 reflections	$\Delta\rho_{\min} = -0.35 \text{ e \AA}^{-3}$
649 parameters	
H-atom parameters constrained	

**Table 1**

Selected geometric parameters ( $\text{\AA}, ^\circ$ ).

S1–C1	1.731 (3)	S3–C25	1.726 (3)
S1–C8	1.810 (3)	S3–C32	1.810 (3)
S2–C18	1.736 (3)	S4–C42	1.736 (3)
S2–C17	1.807 (3)	S4–C41	1.801 (3)
N4–C1–N1	108.4 (2)	N14–C25–N11	108.7 (2)
N4–C1–S1	126.8 (2)	N14–C25–S3	125.8 (2)
N10–C18–N7	108.7 (2)	N20–C42–N17	109.0 (2)
N10–C18–S2	126.9 (2)	N20–C42–S4	126.8 (2)



**Figure 1**

The asymmetric unit of (I), showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 30% probability level.

All H atoms were positioned geometrically and refined as riding (C–H = 0.93–0.97  $\text{\AA}$ ), with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{parent})$ .

Data collection: SMART (Bruker, 1997); cell refinement: SAINT (Bruker, 1997); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1997); software used to prepare material for publication: SHELXTL.

## References

- Bruker (1997). SMART, SAINT (Version 6.22) and SHELXTL (Version 5.10). Bruker AXS Inc., Madison, Wisconsin, USA.
- Bu, X. H., Chen, W., Du, M., Kumar, B., Wang, W. Z. & Zhang, R. H. (2002). *Inorg. Chem.* **41**, 437–439.
- Hong, M. C., Zhao, Y. J., Su, W. P., Cao, R., Fujita, M., Zhou, Z. Y. & Chan, A. S. C. (2000). *Angew. Chem. Int. Ed.* **39**, 2468–2470.
- Juby, P. F., Hudyma, Y. W. & Brown, M. (1968). *J. Med. Chem.* **11**, 111–117.
- Juby, P. F., Hudyma, Y. W., Brown, M., Essery, J. M. & Partyka, R. A. (1982). *J. Med. Chem.* **25**, 1145–1150.
- Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Wang, W., Liu, H.-M. & Zhang, W.-Q. (2004). *Acta Cryst.* **E60**, o1107–o1109.
- Wang, W., Zhao, B., Zheng, P.-W. & Duan, X.-M. (2005). *Acta Cryst.* **E61**, o1163–o1164.
- Zheng, Y., Du, M., Li, J. R., Zhang, R. H. & Bu, X. H. (2003). *J. Chem. Soc. Dalton Trans.* pp. 1509–1514.